

Periodic Table of the Elements

Miss Collins

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American Spellings

hydrogen 1 H 1.00794	beryllium 2 Be 9.012182
lithium 3 Li 6.941	magnesium 12 Mg 24.3050
sodium 11 Na 22.98977	
potassium 19 K 39.0983	calcium 20 Ca 40.078

Alkali metals

Hydrogen
alkaline earth metals

transition metals

poor metals

nonmetals

noble gases

Key:

element name
atomic number
symbol
atomic weight

Rare earth metals

6 7 8 9 10 11 12

3 4 metals

5

boron 5 B 10.811	carbon 6 C 12.0107	nitrogen 7 N 14.00674	oxygen 8 O 15.9994	fluorine 9 F 18.9984
aluminum 13 Al 26.981538	silicon 14 Si 28.0855	phosphorus 15 P 30.97376	sulfur 16 S 32.065	chlorine 17 Cl 35.453
gallium 31 Ga 69.723	germanium 32 Ge 72.64	arsenic 33 As 74.9216	selenium 34 Se 78.96	bromine 35 Br 79.904
tin 50 In 114.818	antimony 51 Sb 118.710	tellurium 52 Te 121.760	iodine 53 I 127.60	xenon 54 Xe 131.293
thallium 81 Tl 204.3833	lead 82 Pb 207.2	bismuth 83 Bi 208.980	polonium 84 Po [209]	astatine 85 At [210]
ununquadium 114 Uuq [289]				radon 86 Rn [222]

3 block

Lanthanides

Actinides

d block

lanthanum
57
La
138.9055

cerium
58
Ce
140.116

praseodymium
59
Pr
140.90765

neodymium
60
Nd
144.24

promethium
61
Pm
[145]

samarium
62
Sm
150.36

europerium
63
Eu
151.964

gadolinium
64
Gd
157.25

terbium
65
Tb
158.9253

dysprosium
66
Dy
162.50

holmium
67
Ho
164.930

erbium
68
Er
167.259

thulium
69
Tm
168.934

ytterbium
70
Yb
173.04

actinium
89
Ac
[227]

thorium
90
Th
232.038

protactinium
91
Pa
231.0359

uranium
92
U
238.0289

neptunium
93
Np
[237]

plutonium
94
Pu
[244]

americium
95
Am
[243]

curium
96
Cm
[247]

berkelium
97
Bk
[247]

californium
98
Cf
[251]

einsteinium
99
Es
[252]

fermium
100
Fm
[257]

mendelevium
101
Md
[258]

nobelium
102
No
[259]

p block

main group transition

Notes: Elements with atomic weights in square brackets have no stable isotopes. Different sources list different atomic weights for elements. The difference arises from the differing atomic weights of various isotopes. We have tried to list the most stable isotope. For example, some sources list the atomic weight of seaborgium as 263 and others 266. The most stable isotope appears to have an atomic weight of 266 so we list that weight here. Roentgenium is still the unofficial name of element 111 but it is the one recommended by the IUPAC so we list it here instead of the generic 'unununium'. Aluminum, cesium, and sulfur are the American spellings for aluminium, caesium, and sulphur. This table was downloaded from http://www.science-teachers.com/printable_periodic_tables.htm